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The Influence of $MoSe_2$ Coated onto Pt Film to DSSC Performance with the Structure TiO_2/Dye/L_xMoSe_2Pt ($0 \le x \le 5$)

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Abstract

Research on the influence of Molybdenum Diselenide ($MoSe_2$) coating on platinum (Pt) to performing dye-sensitized solar cell (DSSC) with the structure of TiO_2/Dye/L_xMoSe_2Pt, ($0 \le x \le 5$) is reported. The hydrothermal method has successfully synthesized the TiO_2 film with square and porous morphology on the indium titanium oxide (ITO) surface. Four peaks of the Raman Scattering detected from the semiconductor confirm the formation of TiO_2 film. The liquid-phase deposition (LPD) also successfully prepared the Pt film. Onto the prepared Pt, the $MoSe_2$ was coated to produce L_xMoSe_2Pt ($0 \le x \le 5$) and then use them as the counter electrode (CE). The best DSSC devices with TiO_2/Dye/L_2MoSe_2Pt structures have resulted in current-density, Voc, and solar cell performance of 11.204 mA/cm^2, 0.66 V, and 2.967%, respectively. The Bode graph confirmed this device has the longest lifetime, proven by the highest peak rise in the lowest frequency. Besides, high-frequency also shows the device has low resistance, useful for accelerating the electrons flow and enhancing DSSC performance.

Keywords: Counter-electrode, semiconductors, L_xMoSe_2Pt, $MoSe_2$, DSSC-performance

Introduction

DSSC is a solar device with semiconductor-based which determined by the effectiveness of the photo-physiochemical phenomenon on the semiconductor [1]. As a third-generation solar-cell has several advantages, such as cheaper, simple-production, and high-efficiency [2]. We believe the efficiency might be enhanced by modifying the key components of the DSSC, such as semiconductor, CE, and electrolyte. Semiconductors which have wide band-gap energy such as TiO_2 [3] with the sense of dye-sensitized material, and the CE with high electro-catalytic, good conductivity and stable [2] could further improve the DSSC performance.

Pt is a popular CE in DSSC application due to it has high efficiency [4, 5] compared to Carbon and Graphene [6-18]. However, an effort to further enhance the Pt properties by adding or attach them to other materials has been reported . For instance, Graphene (G-Pt) and reduced graphene oxide/rGO (rGO-Pt), resulting in the efficiency increased around 0.8% and 0.9%, respectively [19]. Besides, Gong et al. added one-layer of Pt on Graphene (GNS) [6], resulting in the efficiency increased from 4.76% (Pt) to 6.09% (GNS/Pt). Besides, Cheng et al. also reported the MoS_2 addition (Pt/MoS_2) producing the device performance increase of 0.6% [20]. In this

research, we report the use of $MoSe_2$ as a coated layer on platinum and then used them as a CE on the DSSC device. Since it has a very active electrocatalytic, good conductivity [21] and resistant to corrosion caused by electrolytes [22]. The best performance obtained is 2.967% with the V_{OC} and fill-factor generated by 0.66 V and 40.11%, respectively.

Material and Methods

Synthesis TiO_2 and Pt + $MoSe_2$ (L_xMoSe_2Pt, $0 \leq x \leq 5$)

We purchased all chemicals in this work from Sigma-Aldrich. We synthesized TiO_2 semiconductor on the ITO substrate by using a growth solution which consists of 5 mL Ammonium-Hexafluorotitanate and 5 mL boric acid with deionized (DI) water. The detail of the synthesis process has been well-explained in the previous report [3]. Pt films were synthesized using the LPD method on the ITO substrate. We started the Pt synthesis from seeding 3 times by using a solution of L-Ascorbic Acid and potassium tetrachloroplatinate (K_2PtCl_4) with a temperature of 50°C for 2 hours. After that, we continued to Pt growth, using a solution of K_2PtCl_4, L-Ascorbic Acid, Polyvinyl pyrrolidone, and natrium hydroxide with a temperature of 50°C for 5 hours. Last, the prepared Pt was then annealed in an oven for 1 hour at 250°C. Next, the resultant Pt was coated by $MoSe_2$, forming the L_xMoSe_2Pt ($0 \leq x \leq 5$) film using the LPD method. The Pt film was put into the growth-solution (0.5 M of Hexamethylenetetramine 0.5 M, 0.05 M of Ammonium Tetrathromolybdate, 0.1 M of Sodium Borohydrate and 0.01 M of Selenium) and synthesized using water-bath at 90°C for 30 minutes. We repeated these steps to produce two until five layers of $MoSe_2$. Last, the coated Pt was annealed by hydrogen flow at 300˚C for 3 hours. During the synthesis process, the morphology of TiO_2 and Platinum film was observed with a FESEM and Raman Scattering as well.

Preparation of DSSC Devices with the structured of TiO_2/Dye/L_xMoSe_2Pt ($0 \leq x \leq 5$)

We designed DSSC devices as Fig. 2B to see the influence of L_xMoSe_2Pt ($0 \leq x \leq 5$) film as a CE on the device performance. The TiO_2 semiconductor as a photoanode is assembled (after immersed in dye solution (0.05 mM N719) at room temperature for 15 hours) with a CE using a metal clamp. A para-film with a circle hole of 0.23 cm^2 was sandwiched between the TiO_2 and the L_xMoSe_2Pt and injected the electrolyte solution. Last, the current-voltage (J-V) curve of the DSSC was obtained using the Gamry instrument under illumination by simulated sunlight with an intensity of 100 mW cm^2 to characterize the device performance.

Results and Discussion

FESEM image shows the synthesized TiO_2 successfully covered the entire ITO surface (see Fig. 1A). The TiO_2 particles have a morphology that resembles a square shape (Fig. 1B, 1.C) with porous structures and varying particle sizes (see Fig.1C and 1D). The porous structure is believed caused by the use of high temperatures during the growth process. This condition is useful to absorb more dye and enhanced DSSC performance [3]. Fig. 1E shows the Raman Scattering spectrum consisting of 4 peaks of 141 cm^{-1}, 393 cm^{-1}, 513 cm^{-1}, and 636 cm^{-1} with one peak having high intensity between 100-800 cm^{-1}. The resultant peaks are comparable with the research results of Tian et al. [23] and confirm the formation of TiO_2.

Fig. 2A is the FESEM image of Pt with asymmetrical structures (particle size of 156 ± 16 nm) and covers the entire ITO surface. Next, the synthesized L_xMoSe_2Pt ($0 \leq x \leq 5$) was used as CE in DSSC devices with the structured TiO_2/Dye/L_xMoSe_2Pt (see Fig 2B). The J-V curve of DSSC and the photovoltaic parameter are described in Fig. 2C and Table 1, respectively. From the Table 1 shows the DSSC performance start from 2.858%, and then decreased to 2.364% when using L_1MoSe_2Pt as CE. We believed, it occurred as an effect of the heating repetition during $MoSe_2$ coating, and causing the Pt structure damaged and reduced film adhesion [24]. Interestingly, the DSSC performance increase when using the $L_2MoSe2Pt$ as CE to 2.967%. In this stage the $MoSe_2$ coating rather thick, and covering the Pt structure from directly exposed to high temperature, improving the surface area and their conductivity [6]. Besides, the increasing of the surface area providing sensitization of dye materials, speed-up the electrolyte redox reaction [25], and producing higher performance [26]. Then, the DSSC performance decrease again with the number of $MoSe_2$ coating layer increased (L_3MoSe_2Pt, L_4MoSe_2Pt, and L_5MoSe_2Pt). Besides the previous reason, since the addition of each layer of $MoSe_2$ does not followed by the annealed process, causing the bond between them still fragile and easily damaged also causing this phenomenon. The best DSSC performance shows 0.11% higher than the first devices and confirming the coated of $MoSe_2$ are successfully catalyst the Pt film [2, 27]. Since the TiO_2, dye-materials [16, 18], and electrolyte-selection are far from being optimized, applying the Pt in optimized device will enhanced the DSSC performance.These results also support by Bode graphs (see Fig. 3) which shows that this device produces the highest peak on the lowest frequency. Its means, high-frequency peaks show the device has small resistance and highest lifetime compared to other [28]. Besides, the small resistance leading to speed-up the flow of electrons [25], and producing the DSSC devices with the higher performance [29].

Conclusion

The influence of $MoSe_2$ coated onto Pt film to produce L_xMoSe_2Pt ($0 \leq x \leq 5$) as CE on the DSSC performance has been carried out. The best DSSC devices with the structure of TiO_2/Dye/L_2MoSe_2Pt have resulted in current density, Voc, and solar cell performance of 11.204 mA/cm^2, 0.66 V, and 2.967%, respectively. Bode graph confirming the device structure has the highest lifetime because of the highest peak is detected on the lowest frequency. Besides, the high-frequency peaks also show small device resistance, leading to accelerating the electrons flow and enhanced DSSC performance.

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Figure Captions

Fig. 1. **A-D)** are the FESEM image of a TiO_2 semiconductor film, and **E)** The Raman Scatering Spectra of TiO_2

Fig. 2. **A)**. The FESEM image of Pt film (magnification of 50,000 times), **B)**. DSSC array with the $MoSe_2$ coating. **C)**. The J-V curve of DSSC devices with the structure TiO_2/Dye/$L_x MoSe_2$Pt ($0 \leq x \leq 5$).

Table 1 The photovoltaic parameters of DSSC device with the structure TiO_2/Dye/$L_x MoSe_2$Pt ($0 \leq x \leq 5$)

Fig. 3 Bode graphics of 6 DSSC devices with structure TiO_2/Dye/$L_x MoSe_2$Pt ($0 \leq x \leq 5$)

Figure
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Figure and Table

The Influence of $MoSe_2$ Coated onto Pt Film to DSSC Performance with the Structure TiO_2/Dye/$L_x MoSe_2$Pt ($0 \leq x \leq 5$)

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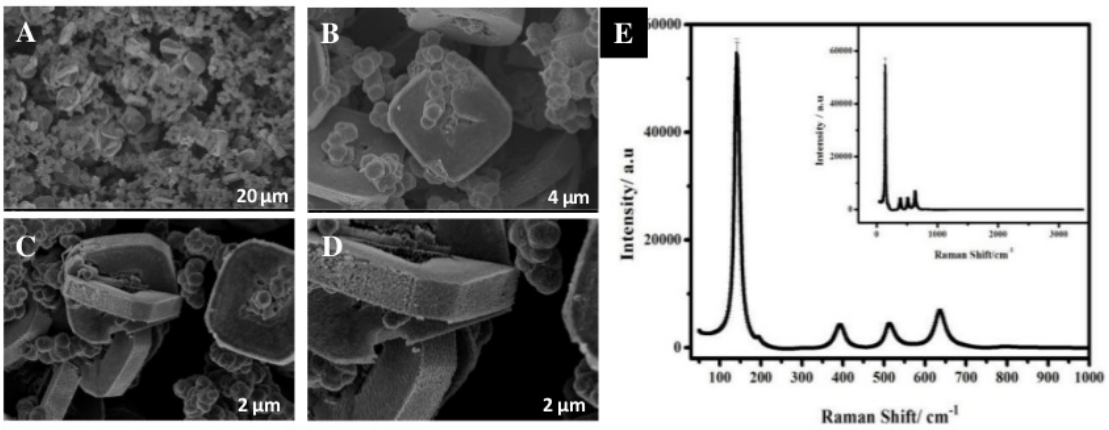

Fig. 1

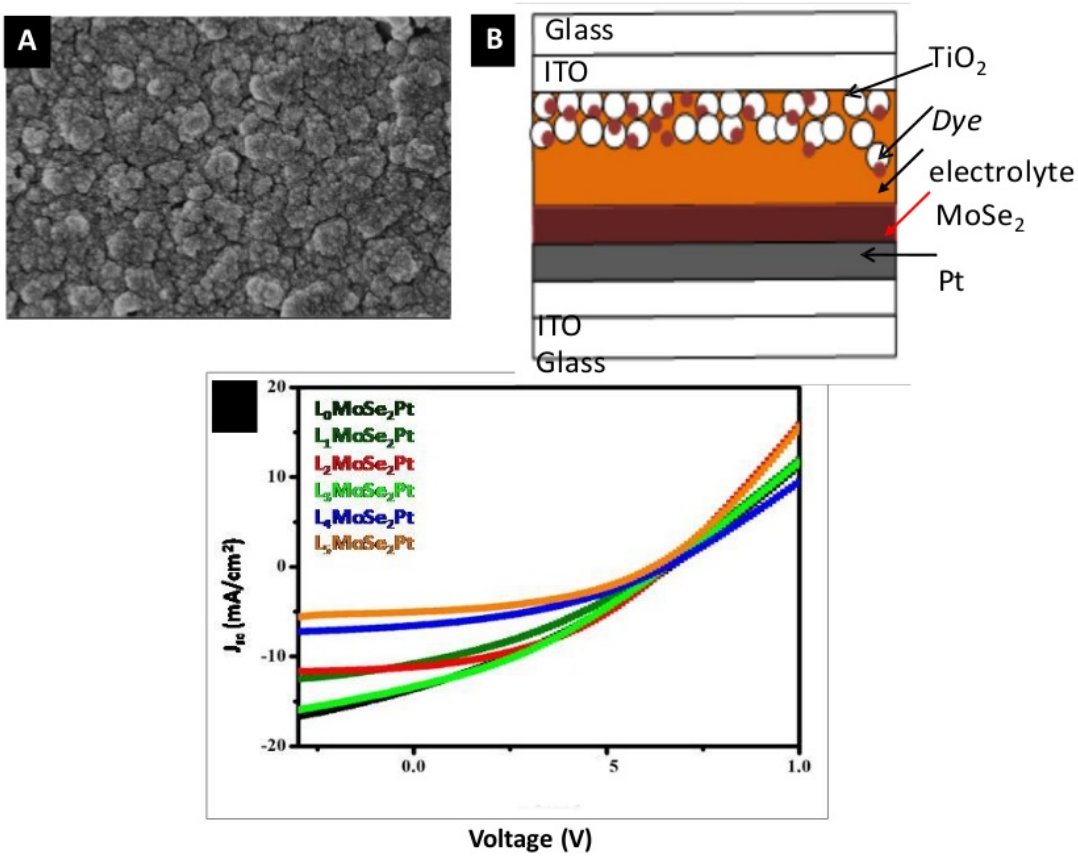
A
B
Glass
ITO
TiO2
Dye
electrolyte
MoSe2
Pt
ITO
Glass
L0MoSe2Pt
L1MoSe2Pt
L2MoSe2Pt
L3MoSe2Pt
L4MoSe2Pt
L5MoSe2Pt
Jsc (mA/cm2)
20
10
0
-10
-20
0.0
5
1.0
Voltage (V)

Fig. 2

Tabel 1

CE	Voc (V)	Jsc (mA/cm^2)	FF(%)	Eff (%)
L_0MoSe_2Pt	0.67	13.635	31.26	2.858
L_1MoSe_2Pt	0.65	10.861	33.47	2.364
$L_2MoSe2Pt$	0.66	11.204	40.11	2.967
L_3MoSe_2Pt	0.66	13.426	32.92	2.921
L_4MoSe_2Pt	0.66	6.600	37.67	1.644
L_5MoSe_2Pt	0.63	5.078	42.13	1.350

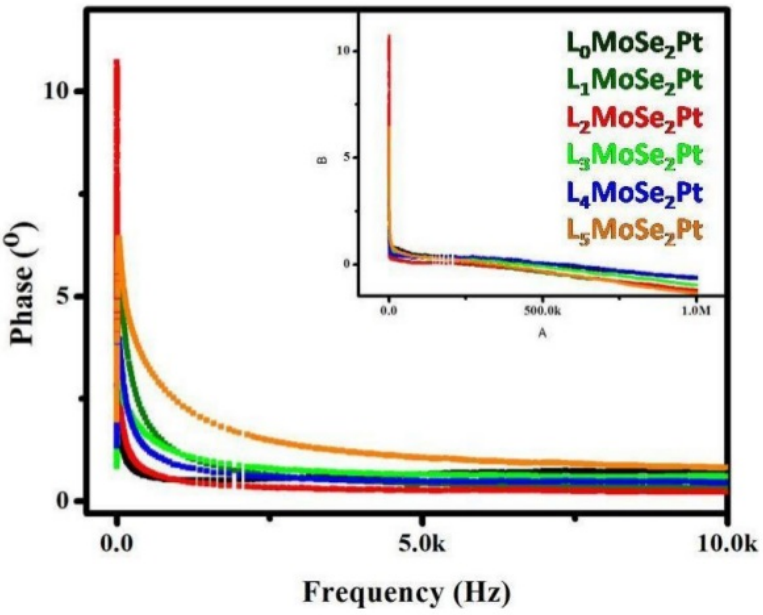
L_0MoSe_2Pt
L_1MoSe_2Pt
L_2MoSe_2Pt
L_3MoSe_2Pt
L_4MoSe_2Pt
L_5MoSe_2Pt
Phase (^{0})
Frequency (Hz)
0.0
5.0k
10.0k
0
5
10
B
A
500.0k
1.0M

Fig 3

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The Influence of $MoSe_2$ Coated onto Pt Film to DSSC Performance with the Structure TiO_2/Dye/$L_x MoSe_2$Pt ($0 \leq x \leq 5$)

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Marjoni Imamora Ali Umar: Data Analysis, Conceptualization, Methodology, Writing- Original draft preparation, Reviewing and Editing.

Resti.: Investigation, Data Collection, Writing- draft preparation.

Venny Haris: Investigation and Validation.

Akrajas Ali Umar*:* Supervision, Conceptualization, Methodology Reviewing and Editing**.**

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*Declaration of Interest Statement

Declaration of interests

☑The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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