

# Plagiasi MI Baru

*by* Marjoni Lensa

---

**Submission date:** 27-May-2020 11:23PM (UTC+0700)

**Submission ID:** 1332873920

**File name:** manuscript\_MLG\_Terbaru\_Bana\_Mei\_2020\_Lensa\_final.docx (1.42M)

**Word count:** 2393

**Character count:** 13278

## Graphene-Au Film Synthesized from GrO in Au-Aquaeus Solution as Counter Electrode For DSSC Application

Marjoni Imamora Ali Umar \*

\*Department of Physics Education, FTIK, Institut Agama Islam Negeri (IAIN) Batusangkar, 27213, West Sumatera, Indonesia

\* Corresponding author Tel.: +62752-71150; fax: +62752-71879.

E-mail: [marjoni.imamora@iainbatusangkar.ac.id](mailto:marjoni.imamora@iainbatusangkar.ac.id)  
[nurjoniimamora@yahoo.com](mailto:nurjoniimamora@yahoo.com)

### Article History

Received:

Reviewed:

Published:

### Key Words

electrical properties;

DSSC Performance;

Au nanoparticles;

four-point probe.

### Abstract

The study on the optical, electrical properties of multilayer graphene (MLG) obtained by thermal-reduction of graphene oxide (GrO) which was synthesized directly by mixing graphite oxide (GO) flake in 0.005, 0.01, 0.015, and 0.02 M of Au aqueous solution has been successfully performed. The resultant GrO was subjected to an annealing temperature of 200°C, 400°C, 500°C for 1h to obtain MLG, and G-Au<sub>2x</sub>, G-Au<sub>4x</sub>, and G-Au<sub>5x</sub> (x=0.005, 0.01, 0.015, and 0.02). The resultant samples were then characterized using FESEM, UV-VIS, four-point probe measurements to study its morphology, optical, and electrical properties. The transmission G-Au increase and its sheet resistant decrease as an increase of annealing temperature. Besides, the annealing treatment was then achieved of its microstructure which is expected may be used as a counter electrode in solar cell applications. The best DSSC devices with Quartz/FTO/ZnO Nanorods/Dye/G-Au<sub>5,01</sub>/Quartz structures have resulted in current-density, Voc, and solar cell performance of 0.1 mA/cm<sup>2</sup>, 0.42 V, and 0.01%, respectively.

### Introduction

Good optical properties are necessary for optoelectronics applications, especially in solar cell devices. Indium titanium oxide and fluoro doped tin-oxide thin film are the famously use as a transparent electrode in optoelectronic applications because they have good transparency and high conductivity at room temperature (Wu et al., 2008). Graphene as a single layer of carbon in 2D which an exceptional property such as electrical (Gilje, Han, Wang, Wang, & Kaner, 2007; G. Wang, Shen, Wang, Yao, & Park, 2009) and optical properties has become a new alternative transparent electrode since founded for almost two decades ago. Nevertheless, it has good transparency but low in conductivity which can be used as an opportunity and challenge. It's peculiar that since it was found that electron mobility in the 2D graphene film exceeding 15.000 m<sup>2</sup>v<sup>-1</sup>s<sup>-1</sup> (G. Wang et al., 2009). Thus graphene providing a

great chance to explore and develop in terms of low-cost preparation (Hong, Xu, Lu, Li, & Shi, 2008), modified composition oxidation agent, and doping materials with enhanced an electrical and transparency of graphene (Eda & Chhowalla, 2010).

Generally, graphite oxide (GO) can be produced either a physics technique or a chemical technique. Among them, a chemical technique, especially by the Hummer's method (William S Hummers Jr & Richard E Offeman, 1958) involving graphite oxidation, exfoliation, and reduction is preferable and a famous technique and yielding important step to produce high-quality properties of graphene film, especially optical and conductivity properties. Graphene can be obtained through vapor exposure of hydrazine (Eda & Chhowalla, 2010; Zhou et al., 2011), thermal-reduction (Choi, Kim, Hwang, Choi, & Jeon, 2011; Kymakis, Stratakis, Stylianakis, Koudoumas, & Fotakis, 2011; Osváth et al., 2007) and combination both of them (Kymakis et al., 2011). Among them, thermal-reduction is very famously applied since the vapor of hydrazine is not sufficient for optimum reduction of the oxide layer (Wu et al., 2008).

Research about the enhanced of optical properties and reduce the sheet resistivity has been reported (Bonaccorso, Sun, Hasan, & Ferrari, 2010), but almost of the obtained that, sheet resistant increase as an increase of the film transmission (X. Li et al., 2009). The present paper will be reported on the tunable of optics, microstructure, and electrical properties as an effect of Au addition during the preparation of graphene oxide film. The resultant properties obtained were compared to the pristine graphene in various annealing treatments with the same composition.

## Experimental

Graphene was prepared from GO which is obtained through modified Hummers-method (W.S. Hummers Jr & R.E. Offeman, 1958). All the materials used in this research such as graphite powder, sodium-nitrate, sulphuric-acid, potassium-permanganate, hydrogen-peroxide, and hydrochloric-chloride were purchased from Sigma-Aldrich. There is 4 main process in this

synthesized namely: oxidation of GO, washing, and filtering and Drying to produce GO flake. The detail of such process has been explained well in a previous report (Umar, Yap, Awang, Jumali, et al., 2013).

To synthesize the GO-Au film, the GO flake was dissolved in 0.005, 0.01, 0.015, 0.02 M of Au<sup>+3</sup> aqueous solution, and in DI at a concentration of 10 mg/ml by stirring and sonication. The resultant solution was spin-coated on quartz substrates by using Spin-Coater. Next, the resulting graphite oxide (GrO) film was put in a furnace and annealed the sample at 200°C, 400°C, and 500°C in argon for 1 h to produce graphene film. The obtained graphene film was labeled as an MLG2, MLG4, MLG5, G-Au2<sub>x</sub>, G-Au4<sub>x</sub>, and G-Au5<sub>x</sub>. The detail of that sample was described in Table 1.

Table 1. The list of sample influence in this research.

No	Concentration	Annealing Temperature (°C)	Sample Name
1	10 mg/ml of DI	200°C	MLG2
		400°C	MLG4
		500°C	MLG5
2	x=0.005, 0.01, 0.015, and 0.02 M of Au <sup>+3</sup>	200°C	G-Au2 <sub>x</sub>
		400°C	G-Au4 <sub>x</sub>
		500°C	G-Au5 <sub>x</sub>

The optical transmittation and morphology of MLG and G-Au films were obtained by using Halo UV-Vis spectrophotometer and Zeiss Supra 55VP FESEM, respectively. The sheet-resistance of the resultant MLG and G-Au film was obtained by Four Point Probe. The current-density-voltage (J-V) of the DSSC was obtained by using the Keithley source meter under illumination by simulated sunlight with an intensity of 100 mW cm<sup>2</sup> to characterize the device performance (See Fig.1).

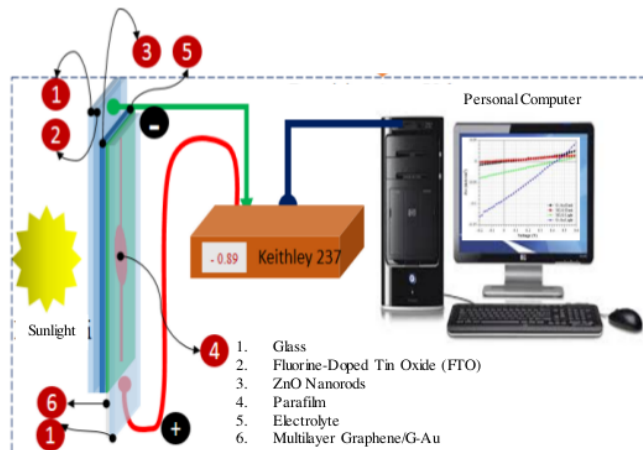


Figure 1. The DSSC array with the structure of Quartz/FTO/ZnO Nanorods/electrolyte/MLG or G-Au/Quartz (Marjoni Imamora, 2014).

## Results and Discussion

Fig. 2 shows that the graph of sheet resistance and optical transmittance at  $\lambda = 550 \text{ nm}$  of G- Au Film at various Au concentration. The complete data for all samples have been summarized in Table 2. From Fig. 2 and Table 2 shows, the G-Au film prepared by using Au solution at 0.01 M succeeded in producing the optimum optical transmittance and lowest sheet resistance film of 68% and  $82 \pm 39 \text{ k}\Omega/\text{sq}$  respectively. MLG-Au has exhibited a good tunable sheet resistivity comparison with the MLG film where GAu5<sub>0.01</sub> which has 82.2 k $\Omega$  of sheet resistance and 68% of its transparency has been shown in Table 2. It's the best sheet resistance in this work and also close to the resultant of reference (Eda et al., 2008). It's believed as an effect of gold nanoparticles addition during GrO preparation which has a unique property with highly dependent on their size, structure, and surrounding (Xiao & Qi, 2011).

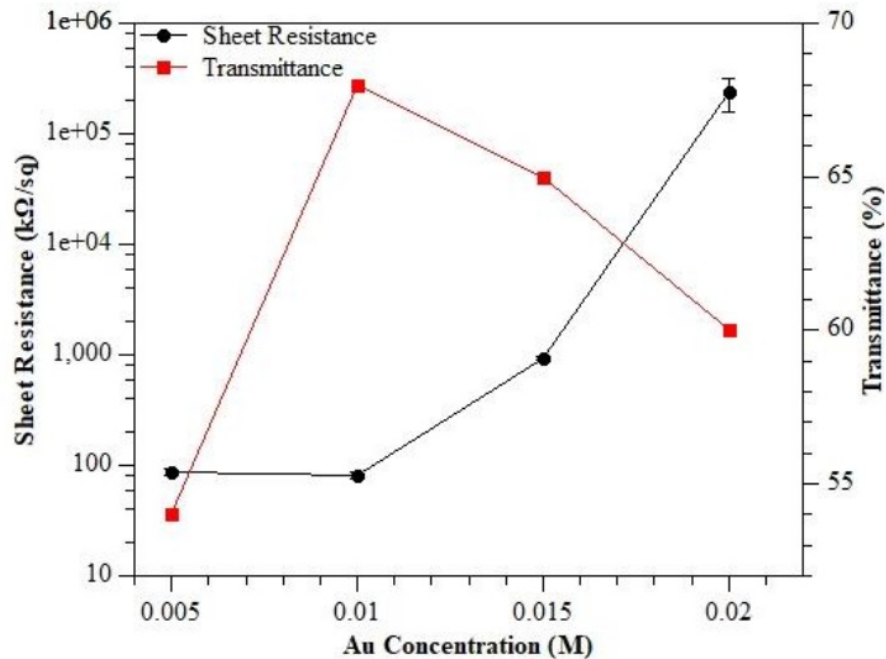


Figure 2. Optical-Transmittance at  $\lambda = 550$  nm versus Sheet-Resistance curve of G-Au Film at Various Au Concentration.

Table 2. The Optical-Transmittance and Sheet-Resistance Data of MLG and G-Au Film Prepared at Various Au Concentration.

Sample	Concentration (M)	Optical Transmittance (T) (%)	Sheet Resistance (SR) kΩ/Sq.	T/SR	Ref
MLG	10	69.2	291.95 ± 63.07	0.24	(Umar, Yap, Awang, Salleh, & Yahaya, 2013)
G-Au <sub>5x</sub>	0.005	54	86.99 ± 5.90	0.62	(Marjoni Imamora, 2014)
	0.01	68	82.21 ± 5.3	0.83	
	0.015	65	932.02 ± 21.7	0.07	
	0.02	60	235718 ± 75811	0.0003	

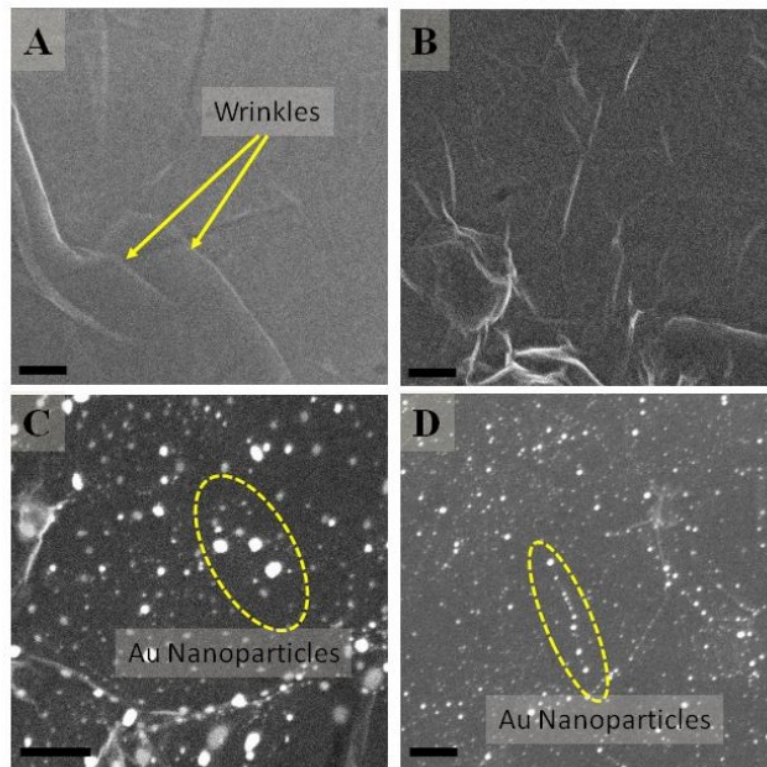


Fig 3. FESEM images of MLG annealing at (A) 400°C (B) 500°C, FESEM images of MLG-Au annealing at (C) 400°C, (D) 500°C (scale bar for (A), (B), (D), and (E) in 200 nm).

Fig. 3 shows the micrograph of MLG4, MLG5, G-Au<sub>4.0.01</sub>, and GAu<sub>5.0.01</sub> film obtained by FESEM characterization. The wrinkles in Fig. 3A occurred as a result of the functional layer already separated during the annealing process. Besides, the increasing of the annealing temperature, graphene aggregation effect, and attractive force between layer during thermal reduction (Cuong et al., 2010) also causing the wrinkles/folds to occur in the film. This phenomenon will make an increasing attenuation coefficient (Xinming Li et al., 2010). It's affected by the increase of the white-light absorbing and maintaining a high surface area of the electrode (C. Wang et al., 2010). The opposite phenomenon occurs, from Fig.3C and 3D show that the microstructure of G-Au. Small particles in Figure 3C and 3D which are circled in yellow-colored are gold particles that successfully attach to multilayer graphene. It's surprisingly improved optics, and microstructure properties through an increasing of annealing

temperature applied. It's believed as an effect of gold nanoparticles addition during GrO preparation which has a unique property with highly dependent on their size, structure, and surrounding (Xiao & Qi, 2011). Besides, it's also led to smoother and brighter G-Au film with decreasing in light absorption. Au particle arrangement is better and flatter (see Fig. 3D) than Fig. 3C. Besides, it has filled in the blank part in between the graphene layer so that Au nanoparticles have been successfully exhibiting to be a potential material in the achieved of optical properties (Treguer-Delapierre, Majimel, Mornet, Duguet, & Ravaine, 2008) of graphene film.

The MLG and G-Au<sub>50.01</sub> as the best sheet-resistance and optical transmittance were used as CE in DSSC. Fig. 4 shows the J-V curves of the DSSC device in dark and under the illumination and the photovoltaic parameter is described in Table 3. The DSSC with MLG film as a CE yielded a short circuit current-density (J<sub>sc</sub>) of 0.03 mA/cm<sup>2</sup>, a Voc of 0.42 V, and a fill-factor (FF) of 25%, resulting in PCE of 0.0027%. On the other hand, the J<sub>sc</sub> and PCE of the DSSC with G-Au<sub>50.01</sub> film as a CE increased significantly which is approximately 300 percent. This result shows that Au nanoparticles were successful to achieve photo-current enhancement in multilayer graphene film (Parvathy Devi, Wu, & Pei, 2011). Besides, the significant increase in J<sub>sc</sub> and PCE could be correlated with the reduction of sheet resistance of the graphene film upon Au addition [23].



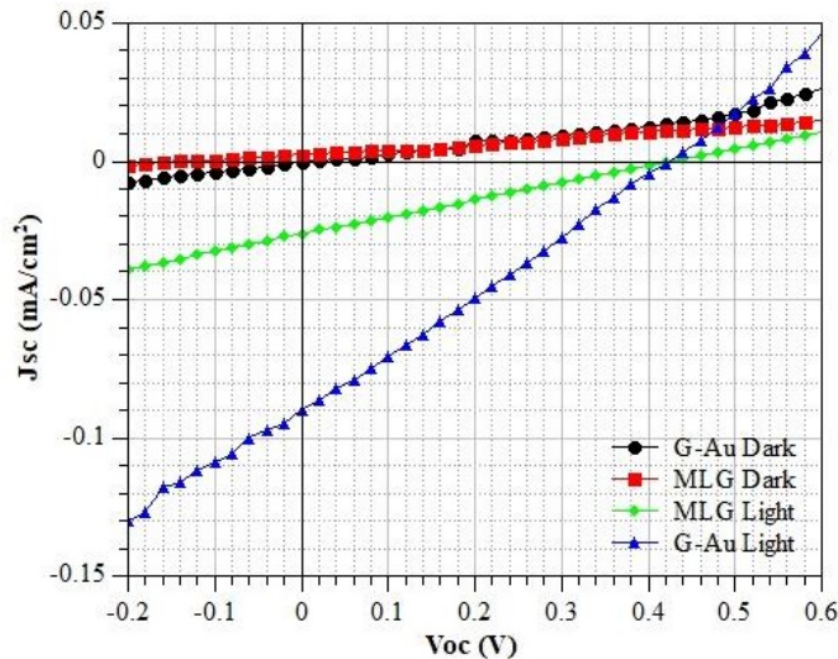


Figure 4. The current-voltage curve of DSSC devices with the structure Quartz/FTO/ZnO Nanorods/Dye/MLG or G-Au/Quartz (Marjoni Imamora, 2014).

Table 3 The photovoltaic parameters data of DSSC with the structure Quartz/FTO/ZnO Nanorods/Dye/MLG or G-Au/Quartz

CE	<sup>3</sup> Voc (V)	Jsc (mA/cm <sup>2</sup> )	FF(%)	Eff (%)
MLG	0.42	0.03	25	0.003
G-Au5	0.42	0.10	26	0.010

## Conclusions

The optical properties of MLG have been successfully improved by using aqueous solution surfactant containing 0.01 M Au since GrO preparation. Herein also the transmission spectra of the G-Au<sub>50.01</sub> sample was then increased linearly with annealing temperature. The microstructure of the G-Au sample was also smoother and flatter as a function of annealing treatment. Amazingly, Au nanoparticles addition during GrO preparation has been

successfully enhanced optical, electrical, and microstructure of resultant graphene. It's making Au-nanoparticle very potential material to enhance the electrical and optical properties.

### Acknowledgments

**1**  
This work has been carried out with the financial support of the Malaysian Ministry of Higher Education (MOHE), under the Research funding ERGS/1/2011/STG/UKM/02/62. The author would like to acknowledge Assoc. Prof Chi Chin Yap, and Dr. Rozidawati Awang who have guided the author.

### References

- Bonaccorso, F., Sun, Z., Hasan, T., & Ferrari, A. C. (2010). Graphene photonics and optoelectronics. [10.1038/nphoton.2010.186]. *Nat Photon*, 4(9), 611-622.
- Choi, H., Kim, H., Hwang, S., Choi, W., & Jeon, M. (2011). Dye-sensitized solar cells using graphene-based carbon nano composite as counter electrode. *Sol Energy Mater Sol Cells.*, 95(1), 323-325. doi: 10.1016/j.solmat.2010.04.044
- Cuong, T. V., Pham, V. H., Tran, Q. T., Chung, J. S., Shin, E. W., Kim, J. S., & Kim, E. J. (2010). Optoelectronic properties of graphene thin films prepared by thermal reduction of graphene oxide. *Materials Letters*, 64(6), 765-767.
- Eda, G., & Chhowalla, M. (2010). Chemically Derived Graphene Oxide: Towards Large Area Thin Film Electronics and Optoelectronics. *Adv Mater*, 22(22), 2392-2415.
- Gilje, S., Han, S., Wang, M., Wang, K. L., & Kaner, R. B. (2007). A chemical route to graphene for device applications. *Nano letters*, 7(11), 3394-3398.
- Hong, W., Xu, Y., Lu, G., Li, C., & Shi, G. (2008). Transparent graphene/PEDOT-PSS composite films as counter electrodes of dye-sensitized solar cells. *Electrochemistry Communications*, 10(10), 1555-1558.
- Hummers Jr, W. S., & Offeman, R. E. (1958). Preparation of graphitic oxide. *Journal of the American Chemical Society*, 80(6), 1339-1339.
- Hummers Jr, W. S., & Offeman, R. E. (1958). Preparation of graphitic oxide. *J Am Chem Soc*, 80(6), 1339.
- Kymakis, E., Stratakis, E., Stylianakis, M., Koudoumas, E., & Fotakis, C. (2011). Spin coated graphene films as the transparent electrode in organic photovoltaic devices. *Thin Solid Films*, 520(4), 1238-1241.
- Li, X., Zhu, H., Wang, K., Wei, J., Fan, G., Li, X., & Wu, D. (2010). Chemical doping and enhanced solar energy conversion of graphene/silicon junctions. *Arxiv preprint arXiv:1012.5730*.
- Li, X., Zhu, Y., Cai, W., Borysiak, M., Han, B., Chen, D., . . . Ruoff, R. S. (2009). Transfer of large-area graphene films for high-performance transparent conductive electrodes. *Nano Lett*, 9(12), 4359-4363.
- Marjoni Imamora. (2014). *Sifat optik dan elektrik filem nipis graphene berbilang lapisan tercampur emas dan tersalut peg*: Universiti Kebangsaan Malaysia.

- Osváth, Z., Darabont, A., Nemes-Incze, P., Horváth, E., Horváth, Z., & Biró, L. (2007). Graphene layers from thermal oxidation of exfoliated graphite plates. *Carbon*, 45(15), 3022-3026.
- Parvathy Devi, B., Wu, K. C., & Pei, Z. (2011). Gold nanomesh induced surface plasmon for photocurrent enhancement in a polymer solar cell. *Solar energy materials and solar cells*.
- Treguer-Delapierre, M., Majimel, J., Mornet, S., Duguet, E., & Ravaine, S. (2008). Synthesis of non-spherical gold nanoparticles. *Gold Bulletin*, 41(2), 195-207.
- Umar, M. I. A., Yap, C. C., Awang, R., Jumali, M. H. H., Salleh, M. M., & Yahaya, M. (2013). Characterization of multilayer graphene prepared from short-time processed graphite oxide flake. *Journal of Materials Science: Materials in Electronics*, 24(4), 1282-1286.
- Umar, M. I. A., Yap, C. C., Awang, R., Salleh, M. M., & Yahaya, M. (2013). *Effect of graphite oxide solution concentration on the properties of multilayer graphene*. Paper presented at the AIP Conference Proceedings.
- Wang, C., Zhang, L., Guo, Z., Xu, J., Wang, H., Zhai, K., & Zhuo, X. (2010). A novel hydrazine electrochemical sensor based on the high specific surface area graphene. *Microchimica Acta*, 169(1), 1-6.
- Wang, G., Shen, X., Wang, B., Yao, J., & Park, J. (2009). Synthesis and characterisation of hydrophilic and organophilic graphene nanosheets. *Carbon*, 47(5), 1359-1364.
- Wu, J., Becerril, H. A., Bao, Z., Liu, Z., Chen, Y., & Peumans, P. (2008). Organic solar cells with solution-processed graphene transparent electrodes. *Appl Phys Lett*, 92, 263302.
- Xiao, J., & Qi, L. (2011). Surfactant-assisted, shape-controlled synthesis of gold nanocrystals. *Nanoscale*, 3(4), 1383-1396.
- Zhou, T., Chen, F., Liu, K., Deng, H., Zhang, Q., Feng, J., & Fu, Q. (2011). A simple and efficient method to prepare graphene by reduction of graphite oxide with sodium hydrosulfite. *Nanotechnology*, 22(045704), 1-6.

# Plagiasi MI Baru

## ORIGINALITY REPORT

9%

SIMILARITY INDEX

6%

INTERNET SOURCES

9%

PUBLICATIONS

%

STUDENT PAPERS

## PRIMARY SOURCES

1

[link.springer.com](https://link.springer.com)

Internet Source

1%

2

Qian Liu. "Polymer Photovoltaic Cells Based on Solution-Processable Graphene and P3HT", *Advanced Functional Materials*, 03/24/2009

Publication

1%

3

[es.scribd.com](https://es.scribd.com)

Internet Source

1%

4

Yaokang Zhang, Sze-Wing Ng, Xi Lu, Zijian Zheng. "Solution-Processed Transparent Electrodes for Emerging Thin-Film Solar Cells", *Chemical Reviews*, 2020

Publication

1%

5

B. Parvathy Devi, Kuo-Cheng Wu, Zingway Pei. "Gold nanomesh induced surface plasmon for photocurrent enhancement in a polymer solar cell", *Solar Energy Materials and Solar Cells*, 2011

Publication

1%

Zhu, Zhaozhao, Trent Mankowski, Kaushik

6

Balakrishnan et al. "Ultra-high aspect ratio copper-nanowire-based hybrid transparent conductive electrodes with PEDOT:PSS and reduced Graphene Oxide exhibiting reduced surface roughness and improved stability", ACS Applied Materials & Interfaces

Publication

<1%

7

[www.ir.nctu.edu.tw](http://www.ir.nctu.edu.tw)

Internet Source

<1%

8

Anuj R. Madaria. "Uniform, highly conductive, and patterned transparent films of a percolating silver nanowire network on rigid and flexible substrates using a dry transfer technique", Nano Research, 07/26/2010

Publication

<1%

9

Jian Cao, Qianyu Liu, Donglai Han, Shuo Yang, Jinghai Yang, Tingting Wang, Haifeng Niu. "One-step hydrothermal synthesis of shape-controlled ZnS–graphene oxide nanocomposites", Journal of Materials Science: Materials in Electronics, 2014

Publication

<1%

10

[ir.lib.nchu.edu.tw](http://ir.lib.nchu.edu.tw)

Internet Source

<1%

11

[iopscience.iop.org](http://iopscience.iop.org)

Internet Source

<1%

12

[olddrji.lbp.world](http://olddrji.lbp.world)

Internet Source

&lt;1%

13

Chayanaphat Chokradjaroen, Ratana Rujiravanit, Sewan Theeramunkong, Nagahiro Saito. "Effect of electrical discharge plasma on cytotoxicity against cancer cells of N,O-carboxymethyl chitosan-stabilized gold nanoparticles", *Carbohydrate Polymers*, 2020

Publication

&lt;1%

14

M. Karimipour, M. Sanjari, M. Molaei. "The synthesis of highly oriented brookite nanosheets using graphene oxide as a sacrificing template", *Journal of Materials Science: Materials in Electronics*, 2017

Publication

&lt;1%

15

P. Swapna, Y. Srinivasa Rao. "Fabrication and characterization of semiconducting single-walled carbon nanotube-based bulk hetero junction organic solar cell using spin coating technique", *Journal of the Chinese Advanced Materials Society*, 2015

Publication

&lt;1%

16

[www.intechopen.com](http://www.intechopen.com)

Internet Source

&lt;1%

17

Luo, Bin, Minghui Liang, Michael Giersig, and Linjie Zhi. "Graphene-Based Materials for Clean

&lt;1%

# Energy Applications", Graphite Graphene and Their Polymer Nanocomposites, 2012.

Publication

---

18

[worldwidescience.org](http://worldwidescience.org)

Internet Source

<1%

---

19

Eric Singh, Hari Singh Nalwa. "Stability of graphene-based heterojunction solar cells", RSC Advances, 2015

Publication

<1%

---

Exclude quotes      On

Exclude matches      Off

Exclude bibliography      On